Convenient syntheses of 2-O-methyl- and 3-O-methyl-D-galactose

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Samples of 2-O-methyl-D-galactose (1) and 3-O-methyl-D-galactose (2) were required as part of a study of the structure of the mycelial wall of *Penicillium* charlesu G. Smith Previously reported syntheses of 1^{2,3} and 2^{3,4} from p-galactose are laborious, or often involve chromatographic purification, at many stages, with low overall yields. An improved procedure for the synthesis of 2 has been described⁵

Benzoylation of benzyl 4,6-O-benzylidene- β -D-galactopyranoside (3) with N-benzoylimidazole has been shown⁶ to give the corresponding 3-benzoate 4 in excellent yield (89-93%). Treatment of 4 with dilute alkali yields the corresponding 2-benzoate 5, also in high yield Diazomethane, with boron trifluoride etherate as a catalyst8, is an effective methylation reagent for partially acylated carbohydrate derivatives that are susceptible to acyl migration under mildly basic conditions. Convenient syntheses of 1 and 2, in good yield from benzoates 4 and 5, using this reagent, are now described.

Treatment of the 3-benzoate 4, in dichloromethane, with excess diazomethane in the presence of boron trifluoride etherate at -5° gave compound 6 as a homogeneous syrup (tlc) This product was debenzoylated with lithium aluminium hydride in tetrahydrofuran to give 74% of benzyl 4,6-O-benzylidene-2-O-methyl- β -D-galactopyranoside (7) Exhaustive hydrogenation of 7 over a palladium catalyst gave crystalline 2-O-methyl-D-galactose (1), which was characterised as 2-O-methyl-N-phenyl-D-galactopyranosylamine

- 7 $R^1 = Me_1 R^2 = H$ 4 $R^1 = H, R^2 = Bz$ 8 $R^1 = Bz, R^2 = Me$ 5 $R^1 = Bz, R^2 = Me$ 9 $R^1 = H, R^2 = Me$

Similar methylation of the 2-benzoate 5 with the diazomethane reagent yielded 84–87% of benzyl 2-O-benzoyl-4,6-O-benzylidene-3-O-methyl-β-D-galactopyranoside (8) This product was debenzoylated with sodium methoxide in methanol to give benzyl 4,6-O-benzylidene-3-O-methyl- β -D-galactopyranoside (9), which, on catalytic 128 NOTE

hydrogenation over palladium oxide, yielded 3-O-methyl- α -D-galactose (2) The product was characterised as 3-O-methyl-D-lyxo-hexose phenylosazone

Methylation of compound 4 with methyl iodide-silver oxide in N,N-dimethylformamide yielded a mixture of four components (t l c), no attempt at separation was made Similar treatment of the 2-benzoate 5 gave compound 8 in 41% yield The product was difficult to purify due to the presence of the isomer 6, presumably formed after migration of the O-2 benzoyl group in 5 to position 3 under the mildly alkaline conditions. It has been shown that, in the interconversion of 4 and 5 under homogeneous conditions, 5 was slightly the more-stable isomer

EXPERIMENTAL

Descending paper chromatography and t1c were performed as described previously⁹ Ir spectra were determined for Nujol mulls Dichloromethane was redistilled from phosphorus pentaoxide before use All melting points are uncorrected

Benzyl 3-O-benzoyl-4,6-O-benzyludene-2-O-methyl-β-D-galactopyranoside (6) — Compound 4^6 (1.5 g) dissolved in dichloromethane (30 ml) at -5° , was treated with boron trifluoride etherate (0 05 ml) and, at the same temperature, diazomethane⁸ in dichloromethane was then added until a faint yellow colour persisted in the solution for 10-15 sec. After 1 h at 0°, polymethylene was filtered off, and the filtrate was washed successively with saturated, aqueous sodium hydrogen carbonate and water, dried (Na₂SO₄), and evaporated to yield 6 as a syrup (1 48 g, 96%), $[\alpha]_D^{24}$ +61 5° (c 1 06, dichloromethane), v_{max} 1715 cm⁻¹ (COPh) (Found C, 69 5, H, 5 5 C₂₈H₂₈O₇ calc · C, 70 6, H, 5 9%)

Benzyl 4,6-O-benzylıdene-2-O-methyl-β-D-galactopyranosıde (7) — A solution of 6 (1 1 g) in dry tetrahydrofuran (40 ml) was added dropwise with stirring to a mixture of lithium aluminium hydride (1 g) in tetrahydrofuran (30 ml), and the mixture was then heated under reflux for 3 h. Cautious addition of water (6 ml) gave a gelatinous precipitate which was separated by filtration and washed well with hot dichloromethane (100 ml). The combined filtrate and washings were concentrated in vacuo, and the residue was recrystallised from propan-1-ol to give 7 (0 64 g, 74%), m. p. $106-110^{\circ}$, $[\alpha]_D^{23}$ —41 2° (c 0 9, dichloromethane) (Found C, 67 6; H, 6 8 C₂₁H₂₄O₆ cale · C, 67 7, H, 6 5%)

2-O-Methyl-D-galactose (1) — Compound 7 (500 mg), in p-dioxane (30 ml), ethanol (30 ml), and glacial acetic acid (5 ml) was hydrogenated exhaustively in the presence of palladium (from 1 g of the oxide). The filtered mixture was concentrated in vacuo, and the residue was recrystallised from ethanol-ether to yield 1 (185 mg, 71%), mp 145-148°, $[\alpha]_D^{23}$ +81 4° (equil, c1, water), R_{GAL} 1 70, lit 2 mp 147-149°, $[\alpha]_D$ +82 6° (equil)

A small portion of the product in methanol was treated with one equivalent of redistilled aniline, and the mixture was heated under reflux for 2 h. Evaporation of the

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solvent and recrystallization of the residue from acetone-light petroleum (60-80°) gave 2-O-methyl-N-phenyl-p-galactopyranosylamine, m p 162-164°, lit ³ m p 165°.

Benzyl 2-O-benzoyl-4,6-O-benzylıdene-3-O-methyl- β -D-galactopyranosıde (8) — A solution of compound 5^7 (2 0 g) in dichloromethane (40 ml) at -5° was treated with diazomethane-boron trifluoride etherate and processed as described for compound 4 The product was recrystallized from propan-2-ol to yield 8 (1 73–1 8 g, 84–87%), mp 168–170°, $[\alpha]_D^{24}$ +42° (c 1, chloroform), v_{max} 1719 cm⁻¹ (COPh) (Found C, 70 3, H, 60 C₂₈H₂₈O₇ calc C 70 6, H, 5 9%)

A solution of compound 5 (1 0 g) in N,N-dimethylformamide (3 ml) and methyl iodide (5 ml) was treated with silver oxide (3 0 g), and the mixture was stirred for 56 h at room temperature in total darkness. The solids were filtered off and washed with chloroform (100 ml), and the combined filtrate and washings were extracted with 5% aqueous sodium cyanide and water, dried (Na₂SO₄), and evaporated The semi-crystalline residue, after five recrystallisations from propan-2-ol, yielded 8 (412 mg, 41%), mp $166-169^{\circ}$, $[\alpha]_D^{23} + 41.6^{\circ}$ (c 1 8, chloroform)

Benzyl 4,6-O-benzyludene-3-O-methyl- β -D-galactopyranoside (9) — A suspension of compound 8 (10g) in methanol (100 ml) was treated with 2m methanolic sodium methoxide (15 ml) at room temperature for 65 h with stirring The homogeneous solution was neutralised with methanol-washed Dowex-50(H⁺) resin, the solid was filtered off and washed with methanol, and the combined filtrate and washings were evaporated The crystalline residue was recrystallised from propan-2-ol-dichloromethane to give 9 (652 mg, 83 5%), mp 182-183°, $[\alpha]_D^{24}$ -44 2° (c 1 13, dichloromethane) (Found C, 67 8, H, 6 8 C₂₁H₂₄O₆ calc · C, 67 7, H, 6 5%).

3-O-Methyl-D-galactose (2) — Compound 9 (500 mg) was hydrogenolysed in the same manner as described for compound 7 The residue crystallized from absolute ethanol, and recrystallization from the same solvent gave 2 (198 mg, 76%), m p 143–146°, $[\alpha]_D^{24}$ +108 2° (equil, c 0 8, water), R_{GAL} 1 55, lit. 4 m p. 146–149°, $[\alpha]_D$ +82 6° (equil)

A solution of compound 2 (21 mg) in 20% aqueous acetic acid (30 ml) was heated for 5 h at 100° with freshly distilled phenylhydrazine (35 mg) in ethanol (02 ml) and sodium metabisulphite (2 mg) The yellow product, which precipitated on cooling, was recrystallised from water (charcoal treatment) to give 3-O-methylp-lyxo-hexose phenylosazone, mp 176–178°, lit 3 mp 176–179°

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REFERENCES

- 1 R A BULMAN AND G J F CHITTENDEN, Unpublished results
- 2 J W H OLDHAM AND D J BELL, J Amer Chem Soc, 60 (1938) 323, D McCreath and F Smith, J Chem Soc, (1939) 387, E Sorkin and T Reichstein, Helv Chim Acta, 28 (1945) 1,

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- J. S Brimacombe, A. M Mofti, and A. K. Al-Radhi, J. Chem. Soc, C, (1971) 1363
- 3 G. G S DUTTON AND Y. TANAKA, Can J. Chem, 40 (1962) 1146
- 4 F. Reber and T. Reichstein, Helv. Chim. Acta, 28 (1945) 1164, H. Bouveng and B Lindberg, Acta Chem Scand, 10 (1956) 1283; M L. Wolfrom, A. Thompson, and M Inatome, J Amer Chem Soc, 79 (1957) 3868; D H. Ball and J K N Jones, J Chem Soc, (1958) 905; A Stoffyn and P. J. Stoffyn, J Org Chem, 32 (1967) 4001.
- 5 E. G. GROS AND I. O. MASTRONARDI, Carbohyd. Res., 10 (1969) 325
- 6 G J. F CHITTENDEN, Carbohyd Res., 16 (1971) 495
- 7 G J F. CHITTENDEN AND J. G BUCHANAN, Carbohyd Res., 11 (1969) 379
- 8 I. O MASTRONARDI, S M. FLEMATTI, J. O DEFERRARI, AND E G GROS, Carbohyd Res., 3 (1966) 177, J O. DEFERRARI, E G GROS, AND I O MASTRONARDI, ibid., 4 (1967) 432, E G. GROS AND I. O MASTRONARDI, ibid., 10 (1969) 318, J. O DEFERRARI, E G GROS, AND I M E THIEL, Methods Carbohyd Chem., 6 (1972) 365.
- 9 G. J. F. CHITTENDEN, Carbohyd Res , 15 (1970) 101